

Recovery Effect of Onion Peel Extract against H₂O₂-Induced Inhibition of Gap-Junctional Intercellular Communication is Mediated through Quercetin

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Abstract: Cellular oxidative damage mediated by reactive oxygen species has been reported to inhibit gap-junctional intercellular communication (GJIC). In turn, the inhibition of GJIC can be attenuated by functional food compounds with antioxidant properties. In this study, we compared the protective effects of onion peel extract (OPE) and onion flesh extract (OFE) on oxidative stress-mediated GJIC inhibition, and investigated the mechanisms of action responsible. OPE restored H₂O₂-induced GJIC inhibition to a higher degree than OFE in WB-F344 rat liver epithelial cells. OPE was found to inhibit H₂O₂-induced phosphorylation of ERK1/2 and Cx43. A radical scavenging assay demonstrated superiority of OPE over OFE, suggesting that the observed effects might be mediated via an antioxidant mechanism. Quercetin is the major compound that is likely to be responsible for the protective effect against H₂O₂-mediated GJIC inhibition. This study suggests that OPE, a material often discarded, may be of value for the future development of functional food products.

Keywords: connexin (Cx) 43, gap-junctional intercellular communication (GJIC), onion peel extract (OPE), quercetin, reactive oxygen species (ROS)

Practical Application: This study demonstrates that onion peel extract (OPE) exhibits a protective effect against the inhibition of gap-junctional intercellular communication (GJIC) mediated by H₂O₂, which is likely to occur via its antioxidant activity. OPE contains significant concentrations of bioactive phenolic compounds. Reductions in oxidative stress can lead to recovery of GJIC, which has been reported to be implicated in the prevention and treatment of cancers. These findings suggest that onion peel, a common waste product, could be used as potential resources for functional food development. Onion peel could be processed into a quercetin-rich powder or a pill for the prevention of cancer and other oxidative stress-related diseases.

Introduction

Gap-junctional intercellular communication (GJIC) involves the exchange of ions and other cytoplasmic molecules between neighboring cells and contributes to the maintenance of homeostatic balance via the regulation of cellular proliferation, differentiation, and growth (Trosko and Ruch 1998). In normal conditions, most cells of the human body have active and functional GJIC (Kumar and Gilula 1996), whereas carcinogenesis, particularly in its formative stages, is strongly associated with GJIC

inhibition (Trosko and Ruch 1998). Studies have shown that GJIC inhibition can be reversed by foods containing high concentrations of polyphenols. Apple extracts and vitamin C have been shown to have protective effects against hydrogen peroxide (H₂O₂)-mediated GJIC inhibition (Lee and others 2004). In addition, kiwifruit cultivars and their active phenolic compounds have been found to reverse H₂O₂-mediated GJIC inhibition in WB-F344 rat liver epithelial (RLE) cells (Lee and others 2010).

The presence of excessive levels of reactive oxygen species (ROS) can cause oxidative damage and result in the dysregulation of GJIC. Activation of MAPK/ERK kinase (MEK) and extracellular signal-regulated protein kinase 1/2 (ERK1/2) by H₂O₂ can induce hyper-phosphorylation of connexin (Cx)43, a major component of transmembrane protein complexes (Lee and others 2004; Lee and others 2010), leading to the inhibition of GJIC. The phosphorylation of Cx43 has been reported to induce conformational changes in membrane-bound proteins that inhibit GJIC (Bruzzone and others 1996; Na and others 2000; Kang and others 2001). While H₂O₂ is known to inhibit GJIC in WB-F344 RLE cells, this inhibition can be reversed by various antioxidants, such as catalase or natural phytochemicals (Kang and others 2008).

The onion is the second most important crop worldwide (Benitez and others 2011). It has been reported that onion peels

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are wasted during processing (Benitez and others 2011; Gawlik-Dziki and others 2013). Benitez and others reported that about 450000 tons of onion waste is produced in Europe (Benitez and others 2011). Since onion wastes are not suitable for other uses, such as fodder or organic fertilizer, these wastes generate huge environmental problems (Benitez and others 2011). In this regard, this study was designed to provide scientific evidence on the application of onion wastes. To this end, we compared health beneficial effects of the onion peel extract (OPE), which was generally discarded, with the onion flesh extract (OFE).

Naturally occurring antioxidants have been shown to exhibit protective effects against carcinogenesis, underlining the role of ROS-mediated oxidative damage during the development of cancers. Onion has been reported to confer health benefits via its antioxidative, anticarcinogenic, and antiinflammatory activities (Slimstad and others 2007). These effects are likely attributable to the presence of polyphenolic compounds such as flavonoids. Recently, a number of clinical and animal studies have demonstrated that onion peel, which contains high concentrations of quercetin, exhibits effects that are beneficial for human health (Jung and others 2011; Lee and others 2011; Lee and others 2012). Indeed, onion peel supplementation has been shown to improve blood

lipid profiles, glucose levels, and blood pressure in male smokers (Lee and others 2011). Supplementation with OPE also regulates the expression of genes associated with cholesterol metabolism, leading to reductions in LDL cholesterol and an increase in HDL cholesterol in rats fed on a high-fat diet (Lee and others 2012). In addition, OPE improves hyperglycemia and insulin resistance by enhancing the expression of the muscular insulin receptor and GLUT4 in high fat diet/streptozotocin-induced diabetic rats (Jung and others 2011).

Flavonoids in OPE could also be acting as direct kinase inhibitors. Recently, a number of studies have reported that various flavonoids in food can act as direct kinase inhibitors (Lee and others 2008; Kim and others 2011; Choi and others 2013). Quercetin, the most abundant flavonoid in OPE, has been reported to directly inhibit Raf and MEK1 (Lee and others 2008), and other flavonoids in OPE, such as rutin and isorhamnetin, have also been reported to inhibit EGFR and MEK1, respectively (Kim and others 2011; Choi and others 2013). These results suggest the possibility that various flavonoids in OPE might be acting as direct kinase inhibitors upstream of the ERK1/2 and Cx43-mediated signaling pathway as well as antioxidants, thereby improving oxidative stress-mediated GJIC inhibition during tumor promotion. Accordingly, we aimed to investigate the effect of OPE and OFE on oxidative stress-mediated GJIC inhibition and the mechanism responsible, using the WB-F344 RLE cell line. In addition, we aimed to identify the major components of OPE or OFE responsible for the protective effect against GJIC.

Materials and Methods

Materials

Acetonitrile, methanol, and trifluoroacetic acid were obtained from Fisher Scientific (Pittsburgh, Pa., U.S.A.). Dimethyl sulfoxide (DMSO), quercetin, rutin, isorhamnetin, quercitrin,

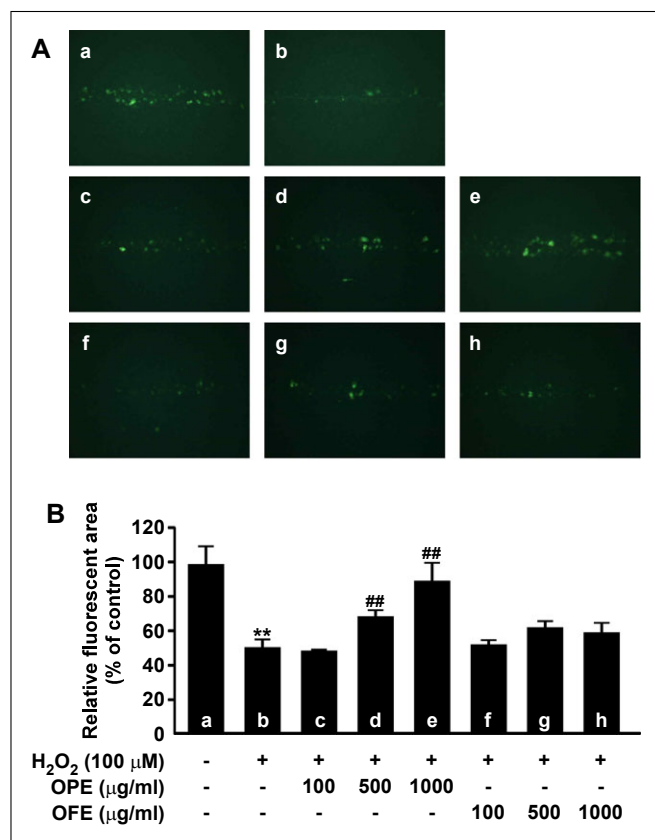


Figure 1—OPE exerts a stronger restorative effect than OFE on H₂O₂-mediated GJIC inhibition in WB-F344 RLE cells. GJIC was assessed using the SL/DT technique, and imaged using an inverted fluorescence microscope (100 \times). Cells were stained with lucifer yellow. (A) Image of GJIC following treatments. (B) Fluorescent area was measured, and the fluorescent area in the untreated control was normalized to 100%. The relative fluorescent area of each group was calculated as a percentage relative to the controls. Data are expressed as means \pm SD values from 3 independent experiments. The asterisks (**) indicate a significant difference between the control group and the group treated with H₂O₂ alone ($P < 0.01$), and the sharps (##) indicate a significant difference between groups treated with H₂O₂ alone and those treated with H₂O₂ plus OPE or OFE ($P < 0.01$).

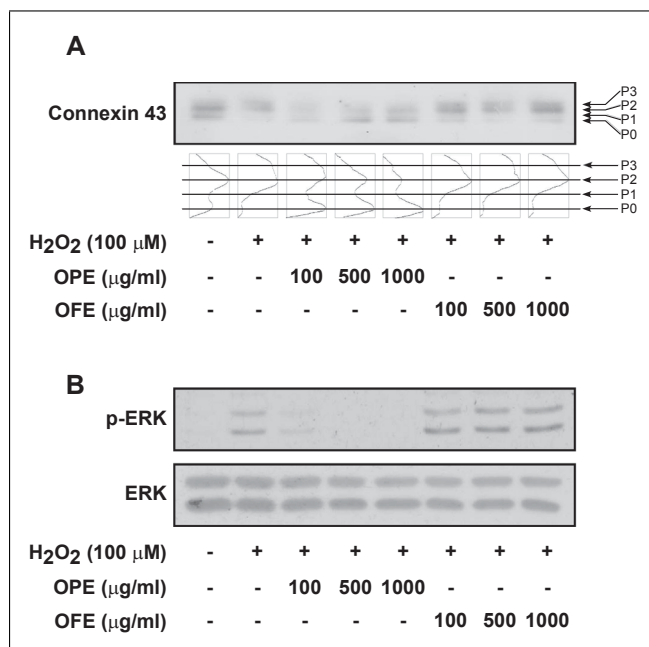


Figure 2—OPE, but not OFE, markedly suppresses H₂O₂-induced phosphorylation of Cx43 and ERK1/2 in WB-F344 RLE cells. (A) The phosphorylation status of Cx43 was evaluated by Western blot analysis. The diagrams show the intensity of Cx43 bands from P0 to P3, which were quantified by using Image J software. (B) The levels of phosphorylated ERK1/2 and total ERK1/2 were evaluated by Western blot analysis.

kaempferol, 1,1-diphenyl-2-picrylhydrazyl, and lucifer yellow were purchased from Sigma-Aldrich (St. Louis, Mo., U.S.A.). 2',7'-Dichlorofluorescein-diacetate was obtained from Invitrogen (Carlsbad, Calif., U.S.A.). The antibodies against phosphorylated ERK1/2 (Tyr202/Tyr204) and total ERK1/2 were purchased from Santa Cruz Biotechnology (Santa Cruz, Calif., U.S.A.), and the antibody against connexin 43 was purchased from Zymed (San Francisco, Calif., U.S.A.).

Preparation of onion peel and onion flesh extracts

Onions (*Allium cepa* L. grown in Jeollanam-do province, Republic of Korea), which were harvested in June 2013, were purchased from a national agricultural and fishery market (Anyang, Republic of Korea). The onions were immediately divided into flesh and peel, which were then freeze-dried (Ilsin, PVTFD 20R, Republic of Korea). Freeze-dried flesh and peel were extracted with 70% (v/v) methanol at room temperature for 24 h, and filtered through filter paper (Whatman Nr. 1). The filtrate was then collected and evaporated (EYELA, NVC-2100, Japan) at 40 °C to remove methanol. The residue was freeze-dried (Ilsin, PVTFD 20R, Republic of Korea) before being powdered with a mortar and pestle, and stored at -20 °C.

Cell culture

The WB-F344 RLE cell line was kindly provided by Dr. J.E. Trosko (Michigan State Univ., East Lansing, Mich., U.S.A.). Cells were cultured in Eagle's modified essential medium (EMEM; Gibco, Gaithersberg, Md., U.S.A.), supplemented with 10% FBS

(Gibco, Gaithersberg, Md., U.S.A.) at 37 °C in a 5% CO₂ humidified incubator (Forma Scientific, Marietta, Ohio, U.S.A.).

GJIC bioassay

GJIC was measured using a scrape-loading/dye-transfer (SL/DT) technique as previously described (el-Fouly and others 1987). Briefly, WB-F344 RLE cells were seeded in 3.5-cm culture dishes (BD Bioscience, Heidelberg, Germany) and incubated until confluence. Cells at confluence were incubated with 100 μM of H₂O₂, supplemented with the onion extracts (OPE or OFE) or individual compounds (quercetin, rutin, and isorhamnetin) for 1 h. Cells were then washed twice with prewarmed phosphate buffer solution (PBS) (2 mL), and Lucifer yellow (1 mg/mL) was added. Cells were then scraped using a scalpel with a surgical-steel blade. After incubation with lucifer yellow solution for 3 min, the cells were washed 3 times with prewarmed PBS (2 mL) and fixed with a 4% formaldehyde solution (2 mL). Fluorescence was observed through an inverted fluorescence microscope at 200× magnification (Olympus, Okaya, Japan), and fluorescence was quantified using Image J software (NIH, Bethesda, Md., U.S.A.).

Western blot analysis

Western blot analysis was performed to assess the phosphorylation of Cx43 and ERK1/2. Cells were seeded and cultured in EMEM supplemented with 10% FBS until 90% confluence. After starvation for 24 h with serum-free EMEM, the cells were treated with onion extracts (OPE or OFE) or individual phenolic compounds (quercetin, rutin, and isorhamnetin) for 30 min, followed

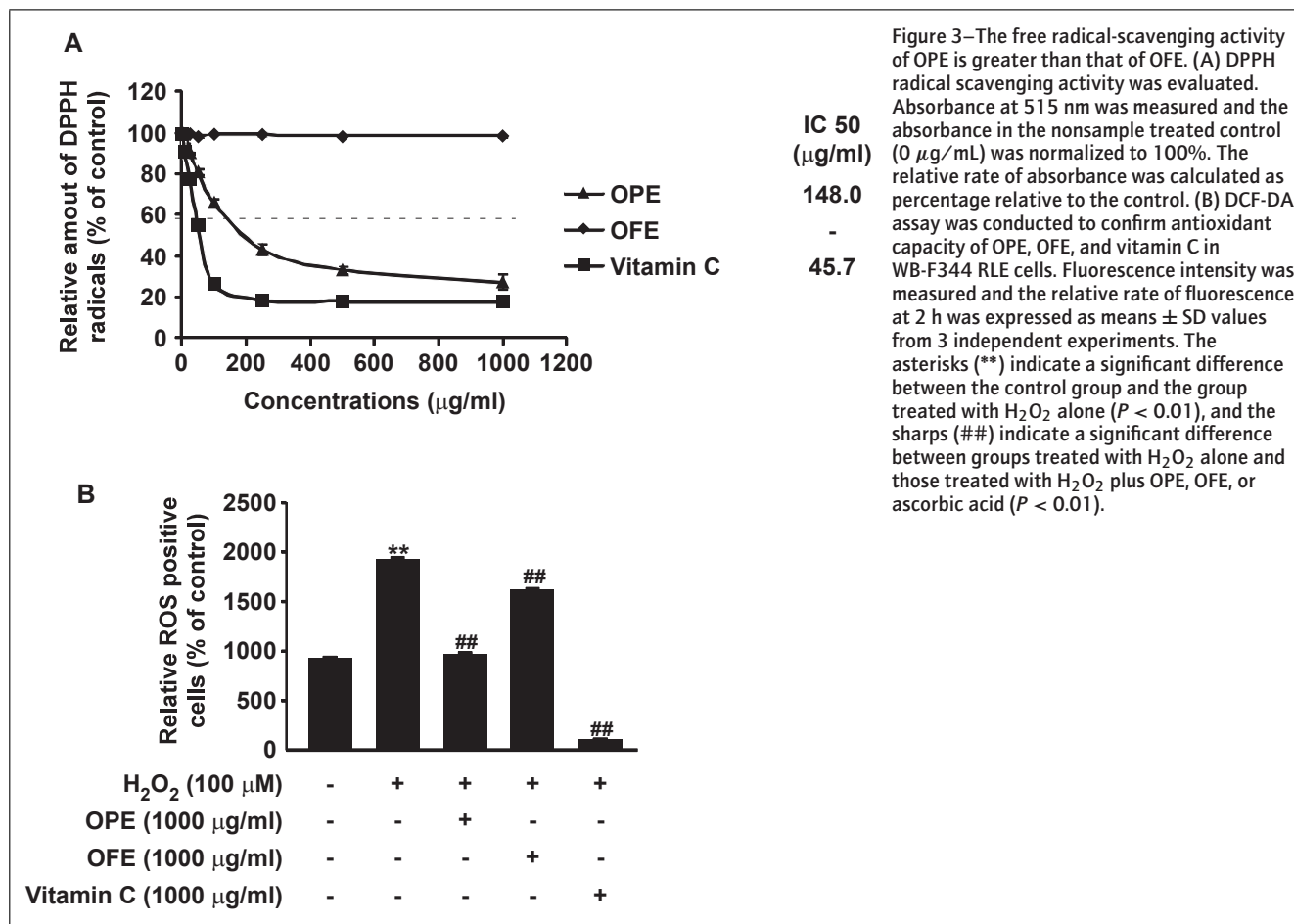


Table 1—Total phenols, total flavonoids, and major polyphenols concentrations in OPE and OFE.

Compounds	OPE	OFE
Total phenols (GAE mg/100 g) ^a	25657 ± 310.3	587 ± 12.0
Total flavonoids (CE mg/100 g) ^b	13214 ± 341.6	72 ± 2.6
Major polyphenols (mg/100 g) ^c	5910 ± 27.7	8.1 ± 0.1
Quercetin (mg/100 g)	4058 ± 33.4	7.7 ± 0.1
Rutin (mg/100 g)	841.5 ± 7.5	ND
Isorhamnetin (mg/100 g)	818.2 ± 3.7	0.4 ± 0.0
Quercitrin (mg/100 g)	103.6 ± 2.1	ND
Kaempferol (mg/100 g)	88.4 ± 0.4	ND

^amg of gallic acid equivalents (GAE) per 100 g fresh weight as standard.

^bmg per catechin equivalents (CE) per 100 g fresh weight as standard.

^cSum of quercetin, rutin, isorhamnetin, quercitrin, kaempferol.

by a treatment with 100 μ M of H₂O₂ for 30 min. The cells were lysed with 1 \times sample buffer (2% SDS, 5% glycerin, 62.5 mM 2-amino-2-(hydroxymethyl) propane-1,3-diol hydrochloride [pH 7.5], 0.05% bromophenol blue, and 10 mM dithiothreitol). Cell lysates were heated at 95 °C for 5 min and centrifuged (14000 rpm, 10 min, 4 °C) to isolate the pellet. Lysates were then separated by sodium dodecylsulfate-polyacrylamide gel electrophoresis (SDS-PAGE) with 12% polyacrylamide gel. Separated proteins were transferred to a 0.45 μ m polyvinylidene fluoride (PVDF) membrane (GE Healthcare, Piscataway, N.J., U.S.A.). The membranes were blocked with 5% skim milk for 1 h and incubated with the indicated primary antibodies followed by horseradish peroxidase

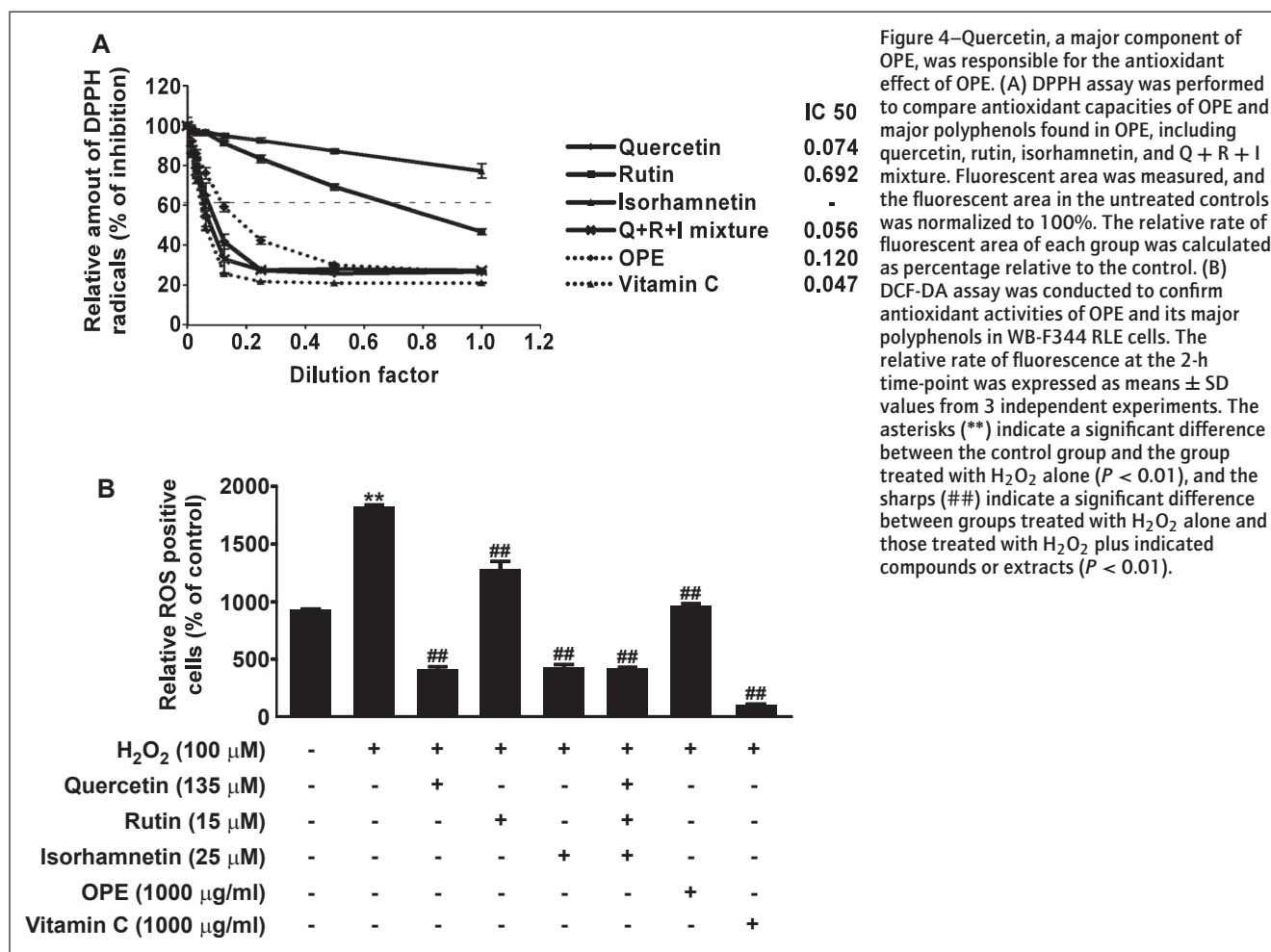
(HRP)-conjugated secondary antibodies. Proteins were visualized using a chemiluminescence detection kit (Amersham Pharmacia Biotech, Piscataway, N.J., U.S.A.).

DPPH assay

1,1-Diphenyl-2-picrylhydrazyl (DPPH) radical-scavenging activity was measured as previously described (Lee and others 2010). DPPH was dissolved in 80% methanol to make 0.1 mM solution. DPPH solution (1993.3 μ L) was mixed with OPE or OFE (6.7 μ L) with various concentrations (0, 10, 25, 50, 100, 250, 500, and 1000 μ g/mL), and loaded into 96-well plates. The mixture was vigorously shaken for 30 min in the dark. The DPPH radical scavenging rate was estimated by measuring absorbance at 515 nm using a spectrophotometer (Hitachi, Tokyo, Japan).

DCF-DA assay

WB-F344 RLE cells were cultured until 90% confluence before trypsinization and centrifugation (2000 rpm, 3 min). Supernatant was removed and the cell pellets were resuspended with 20 μ M of 2',7'-Dichlorofluorescein-diacetate (DCF-DA) in prewarmed PBS. Cells were then treated with the indicated extracts or compounds, followed by the addition of 100 μ M of H₂O₂. Cleavage of DCF-DA was monitored at indicated time points by measuring fluorescence with excitation at 485 nm and emission at 538 nm using a spectrophotometer (Hitachi, Tokyo, Japan).



Quantification of total phenolic and flavonoid contents

The total phenolic content of OPE and OFE were determined using a modified Folin-Ciocalteu method (Singleton and others 1998). Samples (OPE and OFE) or a standard solution was diluted and transferred into a 15-mL tube containing 2.6 mL of deionized water. A reagent blank was also prepared using water. Folin-Ciocalteu phenol reagent (0.2 mL) was added into the mixture. After shaking for 6 min, 7% Na₂CO₃ solution (2 mL) was added, and immediately mixed with deionized water. Following incubation for 90 min at 23 °C, the absorbance relative to that of the prepared blank was read at 750 nm using a spectrophotometer (Model 8453, Agilent Technologies, Palo Alto, Calif., U.S.A.). The total phenolic contents were expressed in mg of gallic acid equivalents (GAE) per 100 g extract weight as a standard curve.

The total flavonoid contents of OPE and OFE were measured using a colorimetric method (Meyers and others 2003). Samples (OPE and OFE) or a standard solution (1 mL) was added to a 15-mL tube containing 4 mL of deionized water. Then 0.3 mL of 5% NaNO₂ was added to the mixture. After incubation for 5 min at 23 °C, 0.3 mL of 10% AlCl₃·6H₂O was added. After incubation for 6 min at 23 °C, 2 mL of 1 M NaOH was added, and the total volume was made up to 10 mL with deionized water. The absorbance of the solution compared with a blank at 510 nm was measured immediately using a spectrophotometer (Model 8453, Agilent Technologies, Palo Alto, Calif., U.S.A.). The total flavonoid contents were expressed as mg of catechin equivalents (CE)/100 g of extract weight as a standard curve.

High-performance liquid chromatography (HPLC) analysis

HPLC analysis of individual compounds was performed as previously described (Crozier and others 1997a, 1997b). Briefly, OPE or OFE (0.2 g) was homogeneously dissolved in 10 mL of methanol. The mixtures were filtered through a 0.45 μm filter (Sartorius, Gottingen, Germany) for HPLC analysis. The mobile phase gradient consisted of water adjusted to pH 2.5 with trifluoroacetic acid (*solution A*) and 100% acetonitrile solution (*Solution B*).

The following gradient was used: 0 to 25 min, 80% to 60% *Solution A*; 25 to 26 min, 60% to 80% *Solution A*; 26 to 30 min, 80% *Solution A*. The operating conditions were as follows: flow rate, 1.0 mL/min; column temperature, 25°C; injection volume, 10 μL; with monitored at 365 nm. HPLC analysis was carried out using 1200 Series HPLC System (Agilent, Palo Alto, Calif., U.S.A.), and analytical separations were performed using an Capcell Pak C18 UG120 column (4.6 mm i.d. × 150 mm, 5 μm, Shiseido, Inc., Tokyo, Japan). The sample concentrations were calculated by comparing the peak areas with external calibration curves.

Statistical analysis

Data are expressed as means ± SD, and Student's *t*-test was used for statistical comparisons. A probability value of *P* < 0.05 was used as the criterion for significance.

Results and Discussion

OPE restores H₂O₂-mediated GJIC inhibition to a greater extent than OFE

GJIC is essential for the maintenance of tissue homeostasis between adjacent cells by exchanging ions, signaling molecules, nucleotides, and other small molecules (Trosko and Ruch 1998). It therefore plays a critical role in cell proliferation, differentiation, and growth of cancer cells (el-Fouly and others 1987). We evaluated the effects of OPE and OFE on H₂O₂-mediated GJIC inhibition using an SL/DT technique with WB-F344 RLE cells. Treatment with H₂O₂ (100 μM) inhibited GJIC by approximately 50% compared to control cells without H₂O₂ treatment (Figure 1A and B). This finding is consistent with studies conducted by others (Crozier and others 1997a; Singleton and others 1998; Rudolf and others 2007). The inhibition was significantly restored by treatments with OPE at 500 and 1000 μg/mL, while OPE at 1000 μg/mL was sufficient to restore GJIC inhibition to a level similar to that of control cells without H₂O₂ treatment (Figure 1A and B). However, treatment of OFE up to 1000 μg/mL

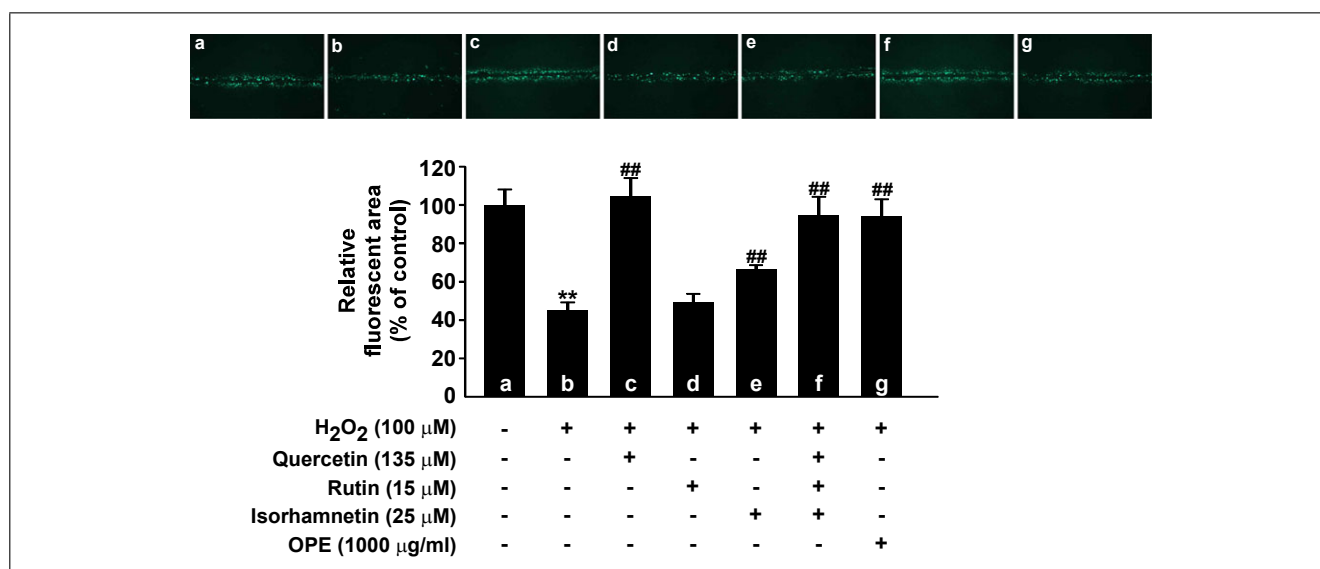


Figure 5—Stronger restorative effect of OPE against H₂O₂-mediated GJIC inhibition was attributed to quercetin in WB-F344 RLE cells. GJIC was assessed using the SL/DT technique, and imaged using an inverted fluorescence microscope (100×). Cells were stained with lucifer yellow. Image of GJIC following treatments, and fluorescent area was measured, and the fluorescent area in untreated controls was normalized to 100%. The relative rate of fluorescent area of each group was calculated as percentage relative to the control. Data are expressed as means ± SD values from 3 independent experiments. The asterisks (**) indicate a significant difference between the control group and the group treated with H₂O₂ alone (*P* < 0.01), and the sharps (##) indicate a significant difference between groups treated with H₂O₂ alone and those treated with H₂O₂ plus OPE or OFE (*P* < 0.01).

did not affect GJIC inhibition induced by H₂O₂ (Figure 1A and B). OPE therefore exhibits a stronger recovery effect against GJIC inhibition induced by oxidative stress than OFE.

OPE suppresses H₂O₂-dependent phosphorylation of Cx43 and ERK1/2

H₂O₂ has been reported to induce phosphorylation of Cx43 and ERK1/2, which has been implicated in the disruption of GJIC (Upham and others 1997; Kang and others 2000). Accordingly, we aimed to investigate whether OPE-mediated GJIC restoration that was normally increased by oxidative stress was associated with decreased Cx43 and ERK1/2 phosphorylation. Antibodies specific for Cx43 were specific for 4 different bands according to phosphorylation status, P0, P1, P2, and P3 (Lee and others 2010). While the most motile band (P0) is indicative of nonphosphorylated Cx43, the least-motile band (P3) represents the most hyper-phosphorylated Cx43. We found that treatment of H₂O₂ (100 μM) causes hyper-phosphorylation of Cx43, evidenced by greater expression of the P3 band compared to P2, P1, and P0 (Figure 2A). Interestingly, treatment with OPE (100, 500, and 1000 μg/mL) dose-dependently shifted the pattern of the bands by enhancing the density of P0 and reducing the density of P3. This suggests that OPE suppresses H₂O₂-induced hyper-phosphorylation of Cx43 (Figure 2A). However, the same amounts of OFE (100, 500, and 1000 μg/mL) did not affect H₂O₂-induced hyper-phosphorylation of Cx43, as seen as a higher density of P3 band (Figure 2A).

Consistent with the finding that OPE only regulates H₂O₂-induced Cx43 phosphorylation, OPE (at 100, 500, and 1000 μg/mL), but not OFE, inhibited ERK1/2 phosphorylation induced by 100 μM of H₂O₂ (Figure 2B). Collectively, these results suggest that OPE, but not OFE, suppresses H₂O₂-induced Cx43 hyper-phosphorylation by inhibiting ERK1/2 phosphorylation, which likely contributes to reverse oxidative stress-mediated GJIC inhibition.

OPE exhibits stronger radical scavenging activity than OFE

We next examined whether the recovery of ROS-mediated GJIC inhibition by OPE was due to its ROS scavenging activity. We measured and compared the radical scavenging activities of OPE and OFE using DPPH and DCF-DA assays. The DPPH radical scavenging activity of OPE and OFE were evaluated and compared with the natural antioxidant vitamin C. We found that OPE (0, 10, 25, 50, 100, 250, 500, and 1000 μg/mL) dose-dependently attenuated DPPH radical formation (Figure 3A). The half maximal inhibitory concentration (IC₅₀) value of OPE was 148 μg/mL, which was higher than that of vitamin C at 45.7 μg/mL. The radical scavenging activity of OPE was not as potent as vitamin C, but stronger than that of OFE. In fact, treatment with OFE at 0, 10, 25, 50, 100, 250, 500, and 1000 μg/mL doses was unaffected by DPPH radical regulation (Figure 3A).

To confirm the intracellular radical scavenging activity of OPE and OFE, a DCF-DA assay was performed using WB-F344 RLE cells. The cells were treated with OPE, OFE, and vitamin C at 1000 μg/mL for 2 h in the presence of H₂O₂ (100 μM). H₂O₂-mediated intracellular ROS formation was approximately 2 times higher than in control cells without H₂O₂ treatment (Figure 3B). Consistent with the finding from the DPPH assay, the intracellular ROS scavenging activity of OPE was greater than OFE. OPE mitigated the formation of ROS by 50% compared to the control cells without H₂O₂ treatment. OFE also attenuated H₂O₂-mediated

ROS formation by 10% compared to H₂O₂-treated control cells (Figure 3B). These results suggest that OPE exhibits a stronger ROS scavenging ability than OFE, and the strong antioxidant capacity of OPE is likely to contribute to the suppression of ERK1/2 phosphorylation and subsequent hyper-phosphorylation of Cx43, thereby protecting against oxidative stress-mediated GJIC inhibition.

OPE contains higher levels of polyphenols and flavonoids than OFE

OPE exhibited a greater antioxidant capacity than OFE, as evident through intracellular ROS scavenging analysis, and subsequently decreased Cx43 phosphorylation and GJIC restoration in H₂O₂-treated WB-F344 RLE cells. We further attempted to evaluate whether GJIC restoration by OPE was attributable to the presence of active polyphenols and flavonoids. We analyzed the total phenolic contents of OPE and OFE using the Folin-Ciocalteu method (Arnous and others 2001). We found that the total levels of polyphenols in OPE was approximately 44 times higher than that in OFE (Table 1). Consistent with the higher levels of total phenolic contents, OPE was found to contain approximately 183 times more total flavonoids than OFE (Table 1). HPLC was then performed to identify the major polyphenols present in OPE and OFE. Quercetin, rutin, isorhamnetin, quercitrin, and kaempferol were identified as being the most abundant in both OPE and OFE. Interestingly, the sum of the total quantity of the 5 polyphenols was 738 times higher in OPE than OFE per unit weight (Table 1). Among these polyphenols, quercetin was found to be the most abundant compound in both OPE and OFE, but the absolute amount of quercetin in OPE was 579 times higher than in OFE. While OFE shows very low levels of other flavonoids, OPE has greater amounts of rutin, isorhamnetin, quercitrin, and kaempferol. However, these flavonoids exist in relatively minor portions compared to quercetin. In summary, quercetin is the most abundant polyphenol found in OPE, which is likely responsible for H₂O₂-induced GJIC inhibition via suppression of Cx43 and ERK1/2 phosphorylation in WB344 RLE cells.

The antioxidant capacity of OPE is attributable to the major polyphenol quercetin

To determine whether the antioxidant capacity of OPE could be attributable to its polyphenol content, we compared the DPPH radical scavenging activity of major OPE components (quercetin, rutin, and isorhamnetin) and a mixture of the 3 compounds to mimic the constituents of OPE. We observed that 1000 μg/mL of OPE contains 135 μM quercetin, 15 μM rutin, and 25 μM isorhamnetin. Therefore, the same concentrations of each polyphenolic component and a mixture of quercetin 135 μM (Q), rutin 15 μM (R), and isorhamnetin 25 μM (I) (referred to as Q + R + I mixture) were used to examine the antioxidant capacity. The DPPH radical scavenging activity was similar between quercetin and the Q + R + I mixture, showing IC₅₀ values of 0.074 and 0.056 μg/mL, respectively. These values were similar to that of vitamin C (0.047 μg/mL), and even lower than that of OPE (0.120 μg/mL) (Figure 4A). In addition, quercetin, isorhamnetin, and the Q + R + I mixture significantly reduced H₂O₂-mediated ROS formation in WB-F344 RLE cells, and even reduced it further than that observed during OPE treatment (Figure 4B). These findings suggest that the antioxidant capacity of OPE is likely attributable to its higher levels of quercetin, and moderate levels of rutin and isorhamnetin.

The DPPH radical scavenging activity of quercetin was greater than the same quantity of OPE, while the DCF-DA assay revealed that the intracellular ROS scavenging activity of isorhamnetin as well as quercetin and the Q + R + I mixture were also greater than that of OPE in WB-F344 RLE cells (Figure 4A and B). The Q + R + I mixture exhibited a greater antioxidant capacity by reducing DPPH radical formation and intracellular ROS formation due to the synergistic effects of quercetin and other antioxidative polyphenols. This suggests that quercetin, isorhamnetin, and the Q + R + I mixture eliminate H₂O₂ and other intracellular free radicals. OPE also reduced intracellular ROS to similar levels as the untreated control cells, but the antioxidant capacity was not as strong as quercetin, isorhamnetin, or the Q + R + I mixture. This could be due to the presence of other minor polyphenols with no antioxidant capacity present in OPE. Many previous studies have reported that an adequate presence of ROS is required for healthy cellular functions, such as maintenance of the cellular innate immune defense system (Conner and others 2002; Leto and Geiszt 2006; Rada and Leto 2008). Therefore, an antioxidant capacity similar to that of control cells without H₂O₂ treatment could be more beneficial for human health than excessive ROS elimination.

Quercetin is responsible for the protective effect of OPE against H₂O₂-induced GJIC inhibition

We further evaluated the effects of OPE, its major polyphenols, and a representative mixture against H₂O₂-mediated GJIC inhibition. OPE, quercetin, and the Q + R + I mixture similarly reversed H₂O₂-mediated GJIC inhibition. Isorhamnetin also significantly restored GJIC inhibition induced by H₂O₂. However, OPE, quercetin, and the Q + R + I mixture restored GJIC inhibition to a greater extent than isorhamnetin (Figure 5). Rutin, one of the major components of OPE, did not reverse GJIC inhibition induced by oxidative stress, which was consistent with an observed weaker antioxidant capacity compared to quercetin and isorhamnetin. Collectively, we found that quercetin is a major factor responsible for the protective effect of OPE against oxidative stress-mediated GJIC inhibition in WB-F344 RLE cells. OPE-mediated GJIC restoration is likely associated with the antioxidant capacity of polyphenols, especially quercetin.

Conclusion

Our results have demonstrated that compared to OFE, OPE exhibits a stronger protective effect against GJIC inhibition by suppressing Cx43 and ERK1/2 phosphorylation during oxidative stress-mediated tumor proliferation, and that this effect is attributable to greater levels of quercetin present in OPE. We conclude that peel extract of onion protects against H₂O₂-mediated GJIC inhibition in WB-F344 RLE cells, suggesting a potential chemopreventive role for oxidative stress-mediated carcinogenesis. This may lead to the development of chemopreventive functional food supplements using onion peel, a waste material that is often discarded during food preparation.

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Disclosure Statement

The authors declared no conflict of interest.

References

- Arnous A, Makris DP, Kefalas P. 2001. Effect of principal polyphenolic components in relation to antioxidant characteristics of aged red wines. *J Agric Food Chem* 49(12):5736–42.
- Benitez V, Molla E, Martin-Cabrejas MA, Aguilera Y, Lopez-Andreu FJ, Cools K, Terry LA, Esteban RM. 2011. Characterization of industrial onion wastes (*Allium cepa* L.): dietary fibre and bioactive compounds. *Plant Foods Human Nutr* 66(1):48–57.
- Bruzzone R, White TW, Paul DL. 1996. Connections with connexins: the molecular basis of direct intercellular signaling. *Eur J Biochem* 238(1):1–27.
- Choi S, Lim TG, Hwang MK, Kim YA, Kim J, Kang NJ, Jang TS, Park JS, Yeom MH, Lee KW. 2013. Rutin inhibits B[a]PDE-induced cyclooxygenase-2 expression by targeting EGFR kinase activity. *Biochem Pharmacol* 86(10):1468–75.
- Conner GE, Salathe M, Forteza R. 2002. Lactoperoxidase and hydrogen peroxide metabolism in the airway. *Am J Respir Crit Care Med* 166(12 Pt 2):S57–61.
- Crozier A, Jensen E, Lean MEJ, McDonald MS. 1997a. Quantitative analysis of flavonoids by reversed-phase high-performance liquid chromatography. *J Chromatogr A* 761(1–2):315–21.
- Crozier A, Lean MEJ, McDonald MS, Black C. 1997b. Quantitative analysis of the flavonoid content of commercial tomatoes, onions, lettuce, and celery. *J Agric Food Chem* 45(3):590–5.
- el-Fouly MH, Trosko JE, Chang CC. 1987. Scrape-loading and dye transfer. A rapid and simple technique to study gap junctional intercellular communication. *Exp Cell Res* 168(2):422–30.
- Gawlik-Dziki U, Swieca M, Dziki D, Baraniak B, Tomilo J, Czyz J. 2013. Quality and antioxidant properties of breads enriched with dry onion (*Allium cepa* L.) skin. *Food Chem* 138(2–3):1621–8.
- Jung JY, Lim Y, Moon MS, Kim JY, Kwon O. 2011. Onion peel extracts ameliorate hyperglycemia and insulin resistance in high fat diet/streptozotocin-induced diabetic rats. *Nutr Metab* 8(1):18.
- Kang KS, Kang BC, Lee BJ, Che JH, Li GX, Trosko JE, Lee YS. 2000. Preventive effect of epicatechin and ginsenoside Rb(2) on the inhibition of gap junctional intercellular communication by TPA and H₂O₂(2). *Cancer Lett* 152(1):97–106.
- Kang KS, Yun JW, Yoon B, Lim YK, Lee YS. 2001. Preventive effect of germanium dioxide on the inhibition of gap junctional intercellular communication by TPA. *Cancer Lett* 166(2):147–53.
- Kang NJ, Lee KM, Kim JH, Lee BK, Kwon JY, Lee KW, Lee HJ. 2008. Inhibition of gap junctional intercellular communication by the green tea polyphenol (-)-epigallocatechin gallate in normal rat liver epithelial cells. *J Agric Food Chem* 56(21):10422–7.
- Kim JE, Lee DE, Lee KW, Son JE, Seo SK, Li J, Jung SK, Heo YS, Mottamal M, Bode AM, Dong Z, Lee HJ. 2011. Isorhamnetin suppresses skin cancer through direct inhibition of MEK1 and PI3-K. *Cancer Prev Res* 4(4):582–91.
- Kumar NM, Gilula NB. 1996. The gap junction communication channel. *Cell* 84(3):381–8.
- Lee KW, Lee SJ, Kang NJ, Lee CY, Lee HJ. 2004. Effects of phenolics in Empire apples on hydrogen peroxide-induced inhibition of gap-junctional intercellular communication. *Biofactors* 21(1–4):361–5.
- Lee KW, Kang NJ, Heo YS, Rogozin EA, Pugliese A, Hwang MK, Bowden GT, Bode AM, Lee HJ, Dong Z. 2008. Raf and MEK protein kinases are direct molecular targets for the chemopreventive effect of quercetin, a major flavonol in red wine. *Cancer Res* 68(3):946–55.
- Lee DE, Shin BJ, Hur HJ, Kim JH, Kim J, Kang NJ, Kim DO, Lee CY, Lee KW, Lee HJ. 2010. Quercetin, the active phenolic component in kiwifruit, prevents hydrogen peroxide-induced inhibition of gap-junction intercellular communication. *Br J Nutr* 104(2):164–70.
- Lee KH, Park E, Lee HJ, Kim MO, Cha YJ, Kim JM, Lee H, Shin MJ. 2011. Effects of daily quercetin-rich supplementation on cardiometabolic risks in male smokers. *Nutr Res Pract* 5(1):28–33.
- Lee SM, Moon J, Do HJ, Chung JH, Lee KH, Cha YJ, Shin MJ. 2012. Onion peel extract increases hepatic low-density lipoprotein receptor and ATP-binding cassette transporter A1 messenger RNA expressions in Sprague-Dawley rats fed a high-fat diet. *Nutr Res* 32(3):210–7.
- Leto TL, Geiszt M. 2006. Role of Nox family NADPH oxidases in host defense. *Antioxid Redox Signal* 8(9–10):1549–61.
- Meyers KJ, Watkins CB, Pritts MP, Liu RH. 2003. Antioxidant and antiproliferative activities of strawberries. *J Agric Food Chem* 51(23):6887–92.
- Na HK, Wilson MR, Kang KS, Chang CC, Grunberger D, Trosko JE. 2000. Restoration of gap junctional intercellular communication by caffeic acid phenethyl ester (CAPE) in a ras-transformed rat liver epithelial cell line. *Cancer Lett* 157(1):31–8.
- Rada B, Leto TL. 2008. Oxidative innate immune defenses by Nox/Duox family NADPH oxidases. *Contrib Microbiol* 15:164–87.
- Rudolf E, Andělová H, Červinka M. 2007. Polyphenolic compounds in chemoprevention of colon cancer – targets and signaling pathways. *Anticancer Agents Med Chem* 7(5):559–75.
- Singleton VL, Orthofer R, Lamuela-Raventós RM. 1998. Analysis of total phenols and other oxidation substrates and antioxidants by means of folin–ciocalteu reagent. *Meth Enzymol* 299:152–78.
- Slimestad R, Fossen T, Vågen IM. 2007. Onions: a source of unique dietary flavonoids. *J Agric Food Chem* 55(25):10067–80.
- Trosko JE, Ruch RJ. 1998. Cell–cell communication in carcinogenesis. *Front Biosci* 3:d208–36.
- Upham BL, Kang KS, Cho HY, Trosko JE. 1997. Hydrogen peroxide inhibits gap junctional intercellular communication in glutathione sufficient but not glutathione deficient cells. *Carcinogenesis* 18(1):37–42.